

## Synthesis and Structures of Steroidal Oxathiolanes

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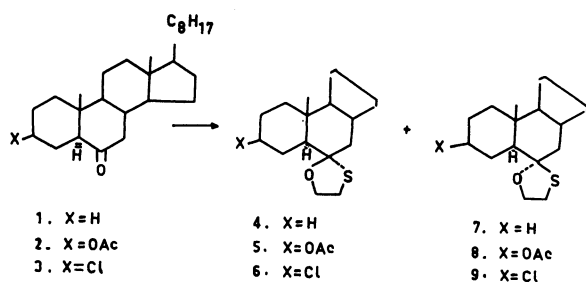
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Synthesis of isomeric oxathiolanes from 6-oxosteroids is described. Configurations at spirocyclic carbon in these compounds have been established on the basis of CD and NMR spectra with the use of shift reagent  $[\text{Eu}(\text{dpm})_3]$ .

As an extension of a previous study,<sup>1)</sup> the synthesis and spectrochemical studies of steroidal oxathiolanes have been carried out.



### Results and Discussion

Reaction of ketone **1** with 2-mercaptoethanol in acetic acid ( $\text{BF}_3 \cdot \text{Et}_2\text{O}$  as catalyst, room temperature) afforded a compound melting at 56 °C and a noncrystallizable oil. These compounds gave molecular ion peaks at  $m/e$  446 ( $\text{C}_{29}\text{H}_{50}\text{OS}$ ). The compound, mp 56 °C, showed characteristic bands at 1225 ( $-\text{S}-\text{CH}_2$ )<sup>2)</sup> and 1065  $\text{cm}^{-1}$  (monothioacetal group),<sup>3)</sup> and the oil bands at 1220 ( $-\text{S}-\text{CH}_2$ ) and 1070  $\text{cm}^{-1}$  (monothioacetal group) in IR spectra. The NMR spectrum of the compound, mp 56 °C, gave two distorted triplets at  $\delta$  4.32 and 3.92 ( $-\text{O}-\text{CH}_2$ ) each integrating for 1 proton, and a double doublets at  $\delta$  2.78 ( $-\text{S}-\text{CH}_2$ ;  $J=3.7$  Hz) for 2 protons. The oil gave a distorted triplet for 2 protons at  $\delta$  4.03 ( $-\text{O}-\text{CH}_2$ ) and a clean triplets for 2 protons at  $\delta$  2.93 ( $-\text{S}-\text{CH}_2$ ).

The configuration of the  $-\text{O}-\text{CH}_2$  and  $-\text{S}-\text{CH}_2$  in

the products could not be assigned with these spectral data, and [tris(dipivalomethanato) europium III] was employed as a shift reagent. Addition of the reagent to the compound, mp 56 °C, caused no significant change in the chemical shift of the signals in its NMR spectrum (Table 1), while its addition to the oily isomer showed a remarkable difference in the chemical shift of the NMR signals (Table 2). This confirmed that the  $-\text{O}-\text{CH}_2$  group of monothioacetal ring attached to C(6) has an equatorial orientation in the oil, and an axial one in the compound, mp 56 °C.<sup>4)</sup> Thus the compound is considered to be (6*S*)-6,6-oxyethylenethio-5 $\alpha$ -cholestane **4** and the oil (6*R*)-6,6-oxyethylenethio-5 $\alpha$ -cholestane **7**. CD data for compounds **4** (negative Cotton effect) and **7** (positive Cotton effect) (Fig. 1) further support configurational assignments for the monothioacetal ring in these compounds.<sup>5)</sup>

By a similar treatment ketones **2** and **3** afforded oxathiolanes **5**, **8**, and **6**, **9** respectively. In NMR spectra, the splitting pattern of  $-\text{O}-\text{CH}_2$  and  $-\text{S}-\text{CH}_2$  protons of **5** and **6** was found identical with **4**, and oxathiolanes **8** and **9** provided similar NMR peaks for  $-\text{O}-\text{CH}_2$  and  $-\text{S}-\text{CH}_2$  protons as in **7**. On the basis of the NMR peak pattern the configurations at spirocyclic carbon in compounds **5**, **6**, and **8**, **9** were assigned similarly to **4** and **7**, respectively. This is supported by the negative Cotton effect (Figs. 2 and 3) for compounds **5** and **6** and positive Cotton effect (Figs. 2 and 3) for compounds **8** and **9**.

The parent ketones **1—3** were generated when monothioacetals **4—9** were treated with aqueous acetic acid.

TABLE 1. INDUCED CHEMICAL SHIFTS ( $\delta$ ) OF VARIOUS PROTONS OF COMPOUND **4** WITH INCREASING AMOUNT OF SHIFT REAGENT  $\text{Eu}(\text{dpm})_3$

	$-\text{O}-\text{CH}_2$	$-\text{S}-\text{CH}_2$	$\text{C}(10)-\text{CH}_3$	$\text{C}(13)-\text{CH}_3$
Sample(20.29 mg) neat	4.05	2.85	0.98	0.66
Sample(20.29 mg) + $\text{Eu}(\text{dpm})_3$ (5.79 mg)	4.00	2.83	0.96	0.66
Sample (20.29 mg) + $\text{Eu}(\text{dpm})_3$ (11.54 mg)	4.06	2.78	1.00	0.67
Sample (20.29 mg) + $\text{Eu}(\text{dpm})_3$ (12.30 mg)	4.15	2.80	0.94	0.61

TABLE 2. INDUCED CHEMICAL SHIFTS ( $\delta$ ) OF VARIOUS PROTONS OF COMPOUND **7** WITH INCREASING AMOUNT OF SHIFT REAGENT  $\text{Eu}(\text{dpm})_3$

	$-\text{O}-\text{CH}_2$	$-\text{S}-\text{CH}_2$	$\text{C}(10)-\text{CH}_3$	$\text{C}(13)-\text{CH}_3$
Sample(22.00 mg) neat	4.03	2.93	0.98	0.67
Sample (22.00 mg) + $\text{Eu}(\text{dpm})_3$ (6.0 mg)	4.20	3.00	0.98	0.67
Sample (22.00 mg) + $\text{Eu}(\text{dpm})_3$ (11.16 mg)	4.63	3.03	1.03	0.70
Sample (22.00 mg) + $\text{Eu}(\text{dpm})_3$ (15.81 mg)	4.91	3.34	1.14	0.72

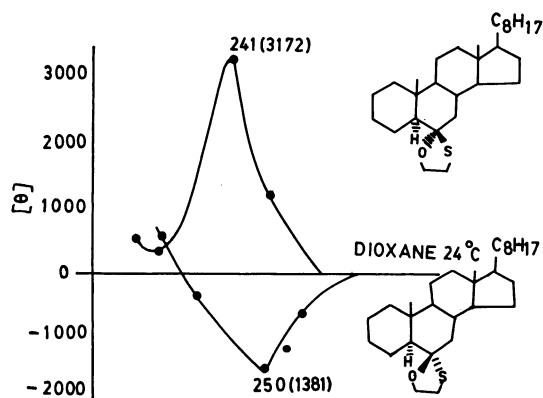


Fig. 1. CD curves of compounds **4** and **7** in dioxane at 24 °C.

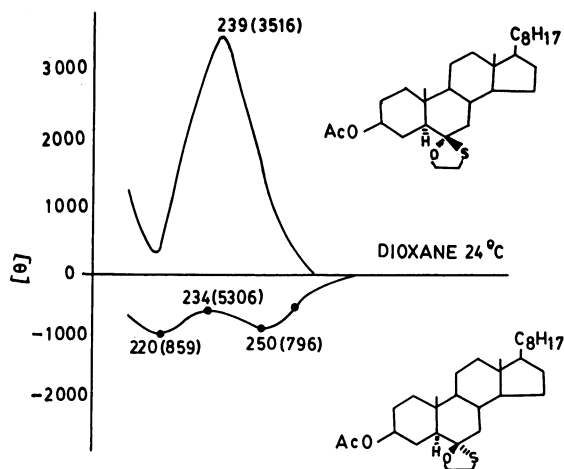


Fig. 2. CD curves of compounds **5** and **8** in dioxane at 24 °C.

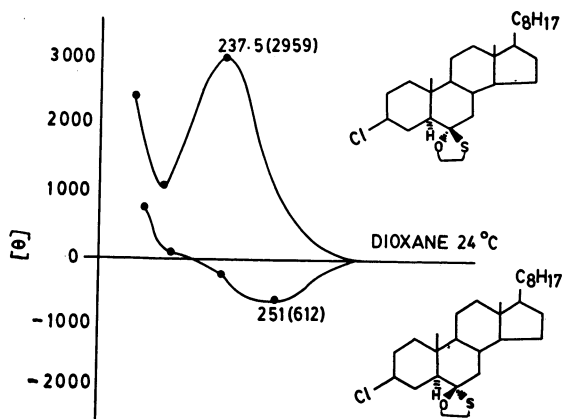


Fig. 3. CD curves of compounds **6** and **9** in dioxane at 24 °C.

### Experimental

All melting points are uncorrected. IR spectra were recorded on a Perkin-Elmer 237 spectrophotometer, and  $^1\text{H}$ -NMR spectra in  $\text{CDCl}_3$  on a Varian A60 instrument ( $\delta$  scale, TMS=0 ppm). CD curves were measured with a JASCO J-20 spectropolarimeter in dioxane. Mass spectra were measured on a Varian AJMS D100 mass spectrometer. TLC were performed with silica gel (BDH) and column chromatography with silica gel (BDH-60-120 mesh). NMR values in

ppm (s=singlet; dd=double doublets; t=triplet; mc=multiplet centred at).

(6S)-6,6-Oxyethylenethio-5 $\alpha$ -cholestane **4** and (6R)-6,6-Oxyethylenethio-5 $\alpha$ -cholestane **7**. A solution of ketone **1**<sup>9</sup> (5.0 g) in AcOH (200 cm<sup>3</sup>) was treated with 2-mercaptoethanol (10 cm<sup>3</sup>) and  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (2 cm<sup>3</sup>) and left to stand at room temperature for 1 h. The solution was diluted with MeOH (25 cm<sup>3</sup>), poured into water and extracted with ether. The ethereal layer was washed successively with water,  $\text{NaHCO}_3$  solution (5%), water and dried (anhydrous  $\text{Na}_2\text{SO}_4$ ). The oil (5.0 g) obtained after removal of ether under reduced pressure was chromatographed over silica gel (100 g). Elution with light petroleum ether afforded **4**; solidified on being left at room temperature (2.0 g); mp 56 °C; MS (70 eV)  $m/e$  446 (30%) ( $\text{M}^+$ ); Found: C, 78.03; H, 11.01%. Calcd for  $\text{C}_{28}\text{H}_{50}\text{OS}$ : C, 78.02; H, 11.20%; CD (dioxane)  $[\theta]^{20}(\text{nm})$ : -1380 (250). NMR ( $\text{CDCl}_3$ ):  $\delta$  0.68, 0.82, 0.92, and 0.98 (C(10)-; C(13)- and side chain methyl protons).

Further elution with light petroleum ether afforded **7** (2.10 g) as an oil; MS (70 eV)  $m/e$  446 (28.3%) ( $\text{M}^+$ ); Found: C, 78.03; H, 11.01%. Calcd for  $\text{C}_{28}\text{H}_{50}\text{OS}$ : C, 78.02; H, 11.20%; CD (dioxane)  $[\theta]^{20}(\text{nm})$ : +3170 (241). NMR ( $\text{CDCl}_3$ ):  $\delta$  0.67, 0.70, 0.75, and 0.98 (C(10)-; C(13)- and side chain methyl protons).

3 $\beta$ -Acetoxy-(6S)-6,6-oxyethylenethio-5 $\alpha$ -cholestane **5** and 3 $\beta$ -Acetoxy-(6R)-6,6-oxyethylenethio-5 $\alpha$ -cholestane **8**. The ketone **2**<sup>7</sup> (5.0 g) was treated with 2-mercaptoethanol (10 cm<sup>3</sup>) and  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (2 cm<sup>3</sup>) in AcOH (200 cm<sup>3</sup>) in the same way as for **1**. The oil obtained after removal of the solvent was chromatographed over silica gel (100 g). Elution with light petroleum ether-ether (22 : 1) afforded **5**, recrystallized from light petroleum ether (2.20 g); mp 142 °C; MS (70 eV)  $m/e$  504 (12.5%) ( $\text{M}^+$ ); Found: C, 73.81; H, 10.49%. Calcd for  $\text{C}_{31}\text{H}_{52}\text{O}_3\text{S}$ : C, 73.8; H, 10.51%; CD (dioxane)  $[\theta]^{20}(\text{nm})$ : -860 (220) and -800 (250). IR (KBr): 1740 ( $\text{CH}_3\text{-CO-O-}$ ), 1240 (C-O), and 1030 cm<sup>-1</sup> (monothioacetal group); NMR ( $\text{CDCl}_3$ ):  $\delta$  4.7 (mc,  $W_{1/2}=16$  Hz, C(3) $\alpha$ -H), 4.28, 3.9 (distorted triplets, -O-CH<sub>2</sub>-), 2.86 (dd,  $J=3.6$  Hz, -S-CH<sub>2</sub>-), 2.03 (s, C(3)-O-CO-CH<sub>3</sub>), 0.70, 0.83, 0.93, and 1.0 (C(10)-, C(13)-, and side chain methyl protons).

Further elution with light petroleum ether-ether (20 : 1) yielded **8** recrystallized from light petroleum ether (2.0 g), mp 106 °C; MS (70 eV)  $m/e$  504 (11.4%) ( $\text{M}^+$ ); Found: C, 73.80; H, 10.49%. Calcd for  $\text{C}_{31}\text{H}_{52}\text{O}_3\text{S}$ : C, 73.8; H, 10.51%; CD (dioxane)  $[\theta]^{20}(\text{nm})$ : +3520 (239). IR (KBr): 1740 ( $\text{CH}_3\text{-CO-O-}$ ), 1240 (C-O), and 1030 cm<sup>-1</sup> (monothioacetal group); NMR ( $\text{CDCl}_3$ ):  $\delta$  4.73 (mc,  $W_{1/2}=16$  Hz, C(3) $\alpha$ -H), 4.03 (distorted triplet, -O-CH<sub>2</sub>-), 1.96 (t, -S-CH<sub>2</sub>-), 2.03 (s, C(3) $\beta$ -O-CO-CH<sub>3</sub>), 0.70, 0.80, 0.90, and 0.98 (C(10)-, C(13)-, and side chain methyl protons).

3 $\beta$ -Chloro-(6S)-6,6-oxyethylenethio-5 $\alpha$ -cholestane **6** and 3 $\beta$ -Chloro-(6R)-6,6-oxyethylenethio-5 $\alpha$ -cholestane **9**. The ketone **3**<sup>8</sup> (5.0 g) was treated with 2-mercaptoethanol (10 cm<sup>3</sup>) and  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (2 cm<sup>3</sup>) in AcOH (200 cm<sup>3</sup>) in the same way as for **1** and **2**. Compounds **6** and **9** were separated. Compound **6** was recrystallized from light petroleum ether (1.8 g), mp 152 °C; MS (70 eV)  $m/e$  480/482 (3 : 1) (26.0%) ( $\text{M}^+$ ); Found: C, 72.30; H, 10.10%. Calcd for  $\text{C}_{28}\text{H}_{48}\text{OSCl}$ : C, 72.5; H, 10.20%; CD (dioxane)  $[\theta]^{20}(\text{nm})$ : -610 (251). IR (KBr): 1070 (monothioacetal group), and 760 cm<sup>-1</sup>; NMR ( $\text{CDCl}_3$ ):  $\delta$  4.0 (mc, C(3) $\alpha$ -H, -O-CH<sub>2</sub>-), 3.86 (dd,  $J=3.6$  Hz, -S-CH<sub>2</sub>-), 0.68, 0.80, 0.92, and 1.0 (C(10)-, C(13)-, and side chain methyl protons).

Compound **9** was recrystallized from petroleum ether (1.75 g); mp 115 °C; MS (70 eV)  $m/e$  480/482 (3 : 1) (7.0%) ( $\text{M}^+$ ); Found: C, 72.31; H, 10.12%. Calcd for  $\text{C}_{28}\text{H}_{48}\text{OSCl}$ : C,

72.5; H, 10.20%; CD (dioxane)  $[\theta]^{20}$  (nm): +2960 (237.5). IR (KBr): 1070 (monothioacetal group), and  $760\text{ cm}^{-1}$  (C-Cl); NMR ( $\text{CDCl}_3$ ):  $\delta$  4.05 (mc, C(3) $\alpha$ -H, -O-CH<sub>2</sub>-), 2.97 (t, -S-CH<sub>2</sub>-), 0.68, 0.78, 0.92, 0.98 (C(10)-, C(13)-, and side chain methyl protons).

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## References

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